

SYNTHESIS OF 5Z,9E-TRIDECADIEN-1-YLACETATE, AN ATTRACTANT OF *Trichoplusia ni*

R. R. Vakhidov* and S. B. Alekseev

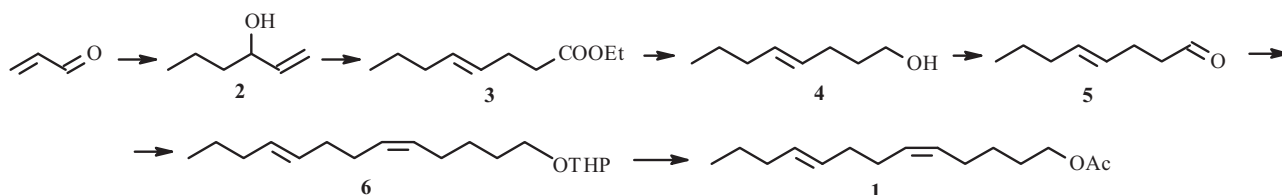
UDC 542.91+547.473+639.936.2

A new synthetic scheme for 5Z,9E-tridecadien-1-ylacetate, an attractant of the cabbage looper Trichoplusia ni (Huebner) was developed using a highly stereoselective Claisen rearrangement.

Keywords: insect pheromones, cabbage looper (*Trichoplusia ni*), 5Z,9E-tridecadien-1-ylacetate, Claisen rearrangement, Wittig reaction.

5Z,9E-Tridecadien-1-ylacetate (**1**) is an attractant of the male cabbage looper (*Trichoplusia ni* Huebner) that affects their mating behavior [1]. Only one synthetic method for this compound that uses the so-called acetylene method has been reported [1].

We developed a new synthetic scheme for **1** that uses as the key step the construction of a double bond with the *E*-configuration via the orthoester version of the Claisen rearrangement.



The starting material in the proposed scheme was 1-hexen-3-ol (**2**), which was readily obtained from reaction of propylmagnesium bromide and acrolein. The Claisen rearrangement occurred easily upon heating allylic alcohol **2** with triethylorthoacetate in the presence of a catalytic amount of propionic acid. This produced the ethyl ester of 4*E*-octenoic acid (**3**). The stereochemical purity of **3** was confirmed by GC and IR and PMR spectral data. In particular, the IR spectrum of **3** exhibited a single absorption band with $\nu = 965 \text{ cm}^{-1}$ in the region of double bond bending vibrations. This was characteristic of *trans* compounds. A band with $\nu = 780 \text{ cm}^{-1}$, characteristic of *cis* compounds, was not observed. Next, **3** was converted to the target molecule using a Wittig reaction through the two intermediates 4*E*-octen-1-ol (**4**) and 4*E*-octenal (**5**). The reaction of **5** with the ylide generated from 1-bromo-5-(tetrahydropyran-2-yl)oxypentane produced 5Z,9E-tridecadien-1-ol tetrahydropyranyl ether (**6**), which was converted in one step to the target compound **1** upon reaction with AcCl–AcOH. The overall yield of **1** was 15.5% calculated per starting acrolein.

EXPERIMENTAL

IR spectra were recorded in thin layers on a Beckman Microlab 620 MX instrument. PMR spectra were taken in $(\text{CD}_3)_2\text{CO}$ solution with HMDS internal standard on a Tesla BS-587A spectrometer (operating frequency 80 MHz). Chemical shifts are given on the δ scale. GC was performed on a Chrom-5 instrument with a flame-ionization detector, column $(3500 \times 3 \text{ mm})$, silicone liquid SE-30 (5%) stationary phase on chromaton N-AW-DMCS $(0.16\text{--}0.20 \text{ mm})$, operating temperature $50\text{--}250^\circ\text{C}$, and He carrier gas (30 mL/min) . TLC was performed on Silufol UV-254 plates with a fixed layer of SiO_2 .

1-Hexen-3-ol (2). The Grignard reagent obtained from propylbromide (24.6 g, 0.2 mol), Mg (4.86 g, 0.2 mol), and anhydrous Et_2O (300 mL) was stirred vigorously, cooled $(-10^\circ \text{ to } -15^\circ\text{C})$, treated dropwise with acrolein (11.2 g, 0.2 mol) in

Ufa State Petroleum Technology University, Salavat Filial, Russia, Salavat, e-mail: vvr2007@mail.ru. Translated from *Khimiya Prirodnikh Soedinenii*, No. 1, pp. 84–85, January–February, 2011. Original article submitted June 16, 2010.

anhydrous Et₂O (40 mL) over 1 h, stirred for 1 h, refluxed for 0.5 h, cooled to 0°C, and decomposed with icewater (70 mL). The aqueous layer was separated and extracted with Et₂O (2 × 50 mL). The combined organic layers were washed with a small quantity of saturated NaCl solution (30 mL), dried over Na₂SO₄, and evaporated. The residue was distilled to afford **2** (14.2 g, 71%), bp 135–136°C [2]. The IR and PMR spectra were analogous to those published [3].

4E-Octenoic Acid Ethyl Ester (3). A mixture of **2** (8.42 g, 84.2 mmol), triethylorthoacetate (50.6 g, 289 mmol), and propionic acid (0.2 g) was stirred at 135–137°C for 2 h, collecting the released EtOH through a fractionating column equipped with a condenser. The mixture was cooled to room temperature, treated with Et₂O (150 mL), washed successively with saturated NaHCO₃ and NaCl solutions, dried over MgSO₄, and evaporated. The residue was vacuum distilled to afford **3** (10.88 g, 76%), bp 104°C (10 mm Hg). IR spectrum (ν, cm⁻¹): 1745 (C=O), 1645, 965 (E-CH=CH). PMR spectrum (δ, ppm, J/Hz): 0.86 (3H, t, J = 6, CH₃), 1.13 (3H, t, J = 6, CH₃CH₂O), 1.3–1.5 (2H, m, CH₂), 1.7–2.0 (4H, m, H₂CC=CCH₂), 2.24 (2H, t, J = 2, CH₂COO), 4.02 (2H, q, J = 6, CH₂O), 5.4–5.8 (2H, m, HC=CH) [4].

4E-Octen-1-ol (4). A suspension of LiAlH₄ (1.45 g, 39.6 mmol) in anhydrous Et₂O (100 mL) was stirred (under Ar), cooled (0–5°C), treated dropwise over 1 h with **3** (6.44 g, 37.9 mmol), heated to room temperature, stirred for another 2 h, cooled to 0°C, and treated dropwise over 0.5 h with H₂O (20 mL). The solution was decanted. The residue was washed with Et₂O (20 mL). The combined Et₂O layers were washed with saturated NaCl solution (3 × 50 mL), dried over Na₂SO₄, and evaporated. The residue was vacuum distilled to afford **4** (3.59 g, 74%), bp 108°C (13 mm) [5]. The IR and PMR spectra were analogous to those published [6].

4E-Octenal (5). A suspension of pyridinium chlorochromate (10.74 g, 4.96 mmol) in dry CHCl₃ (100 mL) was treated dropwise over 2 h with **4** (2.54 g, 19.8 mmol), diluted with Et₂O (100 mL), and passed over a column of silica gel (15 cm). The resulting solution was washed successively with saturated NaHCO₃ and NaCl solutions, dried over Na₂SO₄, and evaporated. The residue was chromatographed over a column of silica gel (eluent hexane:Et₂O, 8:2) to afford aldehyde **5** (2.02 g, 81%). The IR and PMR spectra were analogous to those published [6, 7].

1-(Tetrahydropyran-2-yl)oxy-5Z,9E-tridecadiene (6). 1-(Tetrahydropyran-2-yl)oxy-5-bromopentane (1.62 g, 6.45 mmol) (prepared as before [8]) and PPh₃ (1.69 g, 6.45 mmol) were heated at 160°C for 8 h in an ampul under Ar. The contents were cooled, transferred to a flask, treated with anhydrous THF (50 mL), treated carefully under Ar at –30°C with *t*-BuOK (0.72 g, 6.45 mmol), stirred for 0.5 h, cooled to –70°C, treated with **5** (0.83 g, 6.45 mmol), stirred for 3 h, heated to room temperature, left overnight, diluted with hexane (50 mL), filtered through a fritted-glass filter, and evaporated. The residue was chromatographed over a column of silica gel (eluent hexane then hexane:Et₂O, 8:2) to afford **6** (1.08 g, 60%). IR spectrum (ν, cm⁻¹): 1640, 970 (E-CH=CH), 780 (Z-CH=CH). PMR spectrum (δ, ppm, J/Hz): 0.88 (3H, t, J = 6, CH₃), 1.2–1.6 (12H, m, 6 × CH₂), 1.8–2.1 (8H, m, CH₂C=C), 3.6–3.9 (4H, m, CH₂O), 4.83 (1H, m, CH), 5.1–5.6 (4H, m, HC=CH).

5Z,9E-Tridecadien-1-ylacetate (1). A mixture of **6** (0.62 g, 2.21 mmol), AcOH (1.5 mL), and AcCl (0.35 mL) was refluxed for 5 h (TLC monitoring), cooled, treated with icewater, and extracted with Et₂O (3 × 20 mL). The extract was washed successively with saturated NaHCO₃ and NaCl solutions, dried over Na₂SO₄, and evaporated. The residue was chromatographed over a column of silica gel (eluent hexane:Et₂O, 8:2) to afford **1** (0.42 g, 80%). IR spectrum (ν, cm⁻¹): 1750 (C=O), 1640, 970 (E-CH=CH), 780 (Z-CH=CH). PMR spectrum (δ, ppm, J/Hz): 0.88 (3H, t, J = 6, CH₃), 1.2–1.6 (6H, m, 3 × CH₂), 1.8–2.1 (11H, m, CH₃COO, 4 × CH₂C=C), 3.6–3.9 (2H, m, CH₂O), 5.1–5.6 (4H, m, 2 × HC=CH) [1].

REFERENCES

1. D. Warthen and M. Jacobson, *J. Med. Chem.*, **10**, 1190 (1967).
2. D. Mansuy, J. Leclair, M. Fontecave, and P. Dansette, *Tetrahedron*, **40**, 2847 (1984).
3. D. Savoia, E. Tagliavini, C. Trombini, and A. Umami-Konchi, *J. Org. Chem.*, **46**, 5344 (1981).
4. G. P. Boldrini, M. Mengoli, E. Tagliavini, C. Trombini, and A. Umami-Konchi, *Tetrahedron Lett.*, **27**, 4223 (1986).
5. X.-T. Ding, Y. Ge, Z. Teng, and J.-Y. Fan, *Acta Chim. Sin.*, **45**, 1138 (1987).
6. H. J. Bestmann, K. H. Koschutzky, W. Schatzke, J. Suess, and O. Vostrovsky, *Liebigs Ann. Chem.*, 1705 (1981).
7. B. B. Snider and J. Chen, *Tetrahedron Lett.*, **39**, 5697 (1998).
8. R. N. Shakhmaev, R. R. Vakhidov, V. V. Zorin, and V. N. Odinokov, *Khim. Prir. Soedin.*, 240 (2001).